Stereoselective Preparation of (*E*)-Allyl Alcohols via Radical Elimination from anti- γ -Phenylthio- β -nitro Alcohols

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Phenylthio and hydroxymethyl groups may be introduced into nitroalkenes stereoselectively by treatment with benzenethiol and aqueous formaldehyde to give *anti*- γ -phenylthio- β -nitro alcohols, which are converted into (*E*)-allyl alcohols *via* radical elimination induced by Bu₃SnH.

The formation of carbon–carbon double bonds is important and has been widely explored.¹ Radical elimination should provide a new method for preparing carbon–carbon double bonds, because radical reactions often show selectivities which are different from those for ionic reactions.² The following compounds have been employed for alkene synthesis *via* radical elimination induced by Bu₃SnH: *vic* dibromides,³ β -bromosulphides,⁴ β -bromosulphoxides,⁵ β -bromosulphones,⁶ *vic* dixanthates,⁷ β -isocyanoxanthates,⁸ β -nitrosulphides,⁹ β -nitrosulphones,^{10,11} and *vic* dinitro compounds.¹⁰ Unfortunately, most of these eliminations proceed nonstereospecifically.^{3—8} On the other hand, stereospecific rad-

Table 1. Preparation of *anti*-phenylthio- β -nitroalcohols (1) and their conversion into allyl alcohols and their derivatives (4).

R	% Yield ^a of (1)	Anti/syn ^b	R'	% Yield ^a of (4)	E/Z ^c
Me	(1a) 85	87/13	Bz^{d}	(4a) 75	99/1
Et	(1b) 91	93/7	Bz	(4b) 76	95/5
Prn	(1c) 66	86/14	Bz	(4c) 85	99/1
Pri	(1d) 57	87/13	Bz	(4d) 78	99/1
Bun	(1e) 63	87/13	Ac	(4e) 58	99 /1
n-C ₅ H ₁₁	(1f) 89	87/13	Ac	(4f) 79	99/1
Ph	(1g) 78	87/13	Н	(4 g) 78	95/5
PhCH ₂ CH ₂	(1h) 79	91/9	Н	(4h) 80	99/1

^a Isolated yield. ^b Determined by h.p.l.c. ^c Determined by g.l.c. ^d Bz = PhCO.



Scheme 1. Reagents: i, PhSLi (1.5 equiv.), tetrahydrofuran, room temp., 1 h; ii, 37% HCHO (2.0 equiv.), 3 h; iii, AcOH (2 equiv.), -78 °C, 1 h.



ical elimination occurs in the reaction of β -nitrosulphides or β -nitrosulphones with Bu₃SnH,^{9,10} where the nitro group and the sulphur group are eliminated in the *anti* conformation.¹¹ Previously, we reported a convenient method for preparing *anti*- β -nitrosulphides and β -nitroselenides by kinetically controlled protonation.¹² We now report a new method for preparing (*E*)-allyl alcohols, which consists of a one-step preparation of *anti*- γ -phenylthio- β -nitro alcohols from nitroalkenes, and radical elimination induced by Bu₃SnH.

A mixture of lithium benzenethiolate and the nitroalkene in tetrahydrofuran (THF) was stirred at room temperature for 1 h and then 37% aqueous HCHO was added. The resulting solution was stirred at room temperature for 3 h, and then cooled to -78 °C. Acetic acid (2 equiv.) was added to this



Scheme 2. Reagents: i, Bu₃SnH (3.5 equiv.), azoisobutyronitrile (AIBN) (1.0 equiv.), toluene, 110 °C, 30 min.



Scheme 3

solution at -78 °C; the usual work up afforded the γ -phenyl-thio- β -nitro alcohols (1) in good yield with the *anti*-isomer predominating. The results are summarized in Table 1.

Thus, phenylthio and hydroxymethyl groups are introduced into the nitroalkene in one step with *anti*-selectivity, the *anti*/syn ratio being ~9:1. If the reaction was carried out in the conventional way⁹ using the nitroalkene, benzenethiol, 37% aqueous HCHO, and catalytic amounts of base, complex mixtures of products, namely, a 1:1 mixture of *anti*-(1) and *syn*-(1), bis-hydroxymethylated compounds (5), and β -nitrosulphides (6), were obtained. The present diastereoselectivity is presumed to arise in the protonation of (2) at low temperature as shown in Scheme 1. The stereochemistry of (1) was based on ¹H n.m.r. data. For example, the -CHR-SPh signal of *syn*-(1f) (δ 3.69) appeared at lower field than that of *anti*-(1f) (δ 3.51), because of the strong anisotropic effects of the nitro group. This tendency is generally observed for *syn*and *anti*- β -nitrosulphides and β -nitroselenides.¹²†

Compounds (1) and their acylated derivatives (3) were converted into allyl alcohols and their derivatives (4) on treatment with Bu₃SnH (3.5 equiv.) in the presence of azoisobutyronitrile (AIBN) (1 equiv.) in toluene at 110 °C (Scheme 2). The results are summarized in Table 1.

It is noteworthy that pure (E)-allyl alcohols are selectively obtained from *anti*-rich (1) or (3). For example, the elimina-

[†] The 3-H signal of *anti*- and *syn*-2-nitro-3-phenylselenopentane appeared at δ 3.37 and 3.52, respectively. The *anti*-isomer gave pure (Z)-2-nitro-pent-2-ene on thermal elimination of selenenic acid.¹²

tion reaction of *anti*-rich (**3f**) gave almost pure (E)-(**4f**). However, the elimination reaction from a 1:1 mixture of *anti*-(**3f**) and *syn*-(**3f**) results in the formation of an E/Z mixture of (**4f**), the E/Z ratio being about 7:3 (Scheme 3).

These results suggest that the stereoselective formation of (E)-(4) is mainly derived from *anti*-stereospecific radical elimination of *anti*-(1) or *anti*-(3). The isomerization of (Z)-(4) to (E)-(4) occurs to some extent, but this contribution to the stereoselective formation of (E)-(4) is very small.

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